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# Lithium-ion battery aging: Representative EV cycling profiles compared to calendar life

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- Context and objectives
- Calendar & cycling aging experiments
- Aging results
  - Full cells
  - Post-mortem characterization
- Conclusion

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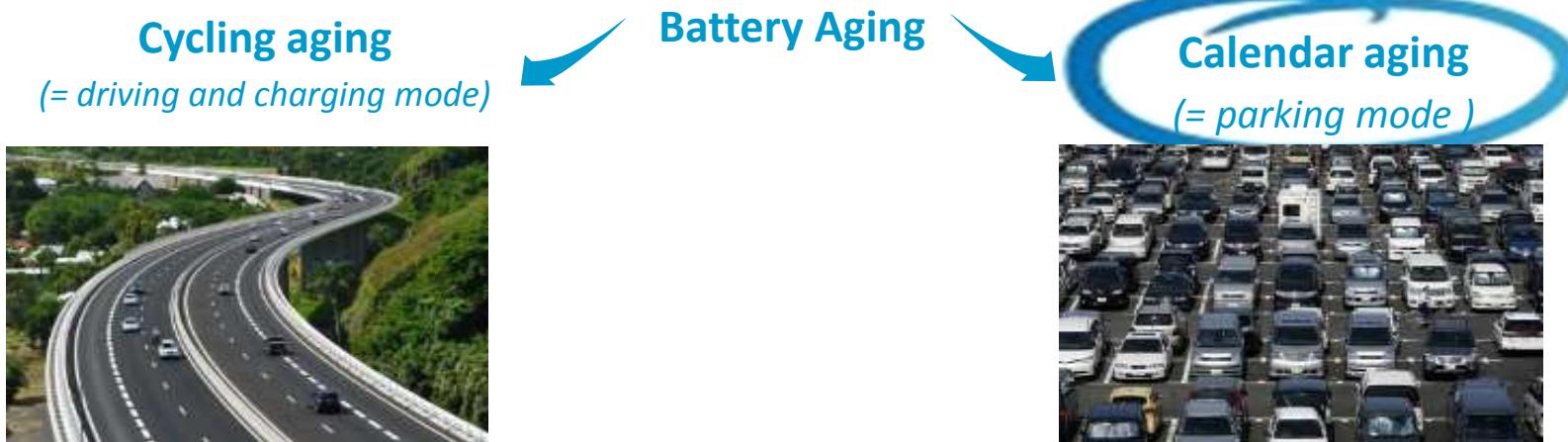
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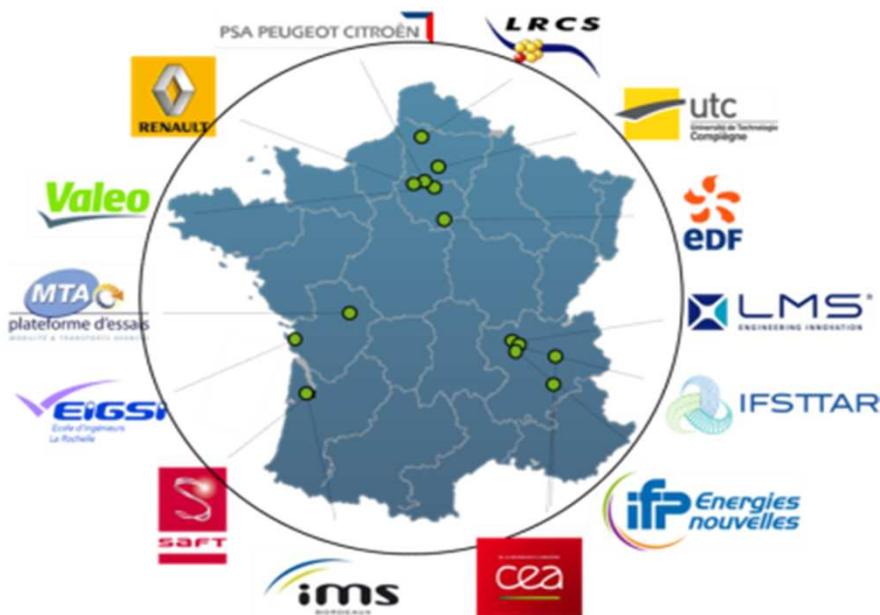
- Lithium-Ion battery : promising candidate for electric transportation
- However performances decline with use and time...



*Passenger vehicles spend about 90% of time in parking mode!*

*10 to 15 years are required !*

- The research project SIMCAL (2010 – 2013) :



- Consortium of 14 partners in both academy and industry
- Calendar aging data for 6 Li-Ion technologies
- In-depth understanding of aging mechanism upon storage (mainly loss of lithium cyclable due to SEI growth)

Question : how cell lifetime is affected by using the battery **once a day** like in a typical automotive application?

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- Aging experiments performed on:
  - Commercial 12 Ah cell (purchased in 2010)
  - Cell chemistry: Li NiCoMnO<sub>2</sub> | Graphite
  - Measured capacity (25°C, 1Cnom):  
 $12.7 \pm 0.12$  Ah

Aging temperature test set to 45°C

1. Calendar aging conditions:

3 SoCs: 30%, 65%, 100%



End-of-discharge voltage (V)	2.70
End-of-charge voltage (V)	<u>4.20</u>



Compared to ...



2. Cycling aging conditions representative of an EV usage

3 charging scenarios investigated

### 1. Calendar tests :

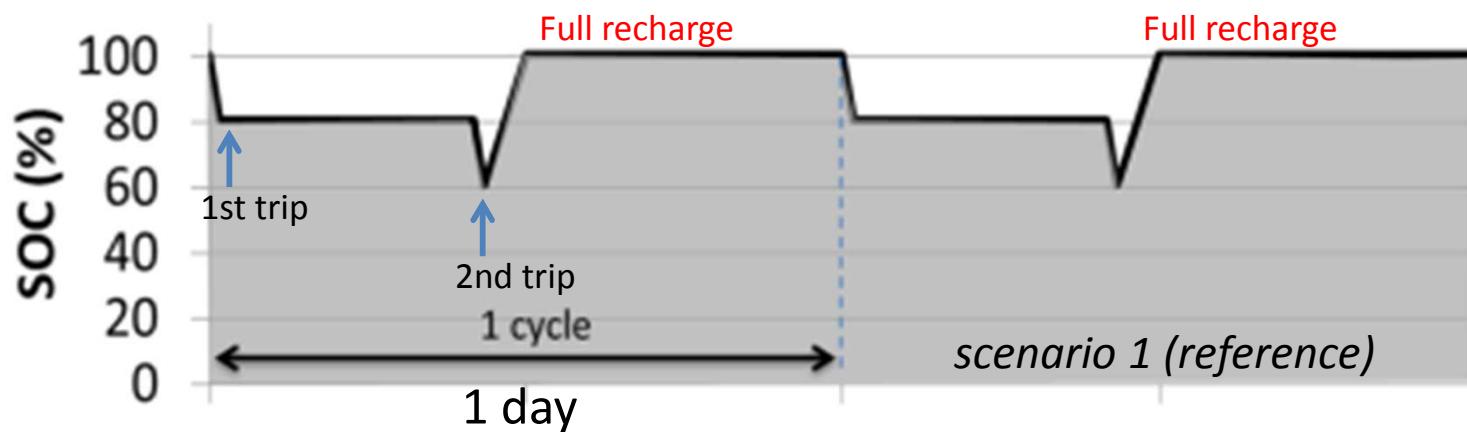
State-of-Charge (%)		
SOC1	SOC2	SOC3
30	65	100



- Open Circuit
- 3 cells for each storage condition
- Duration: 24 months

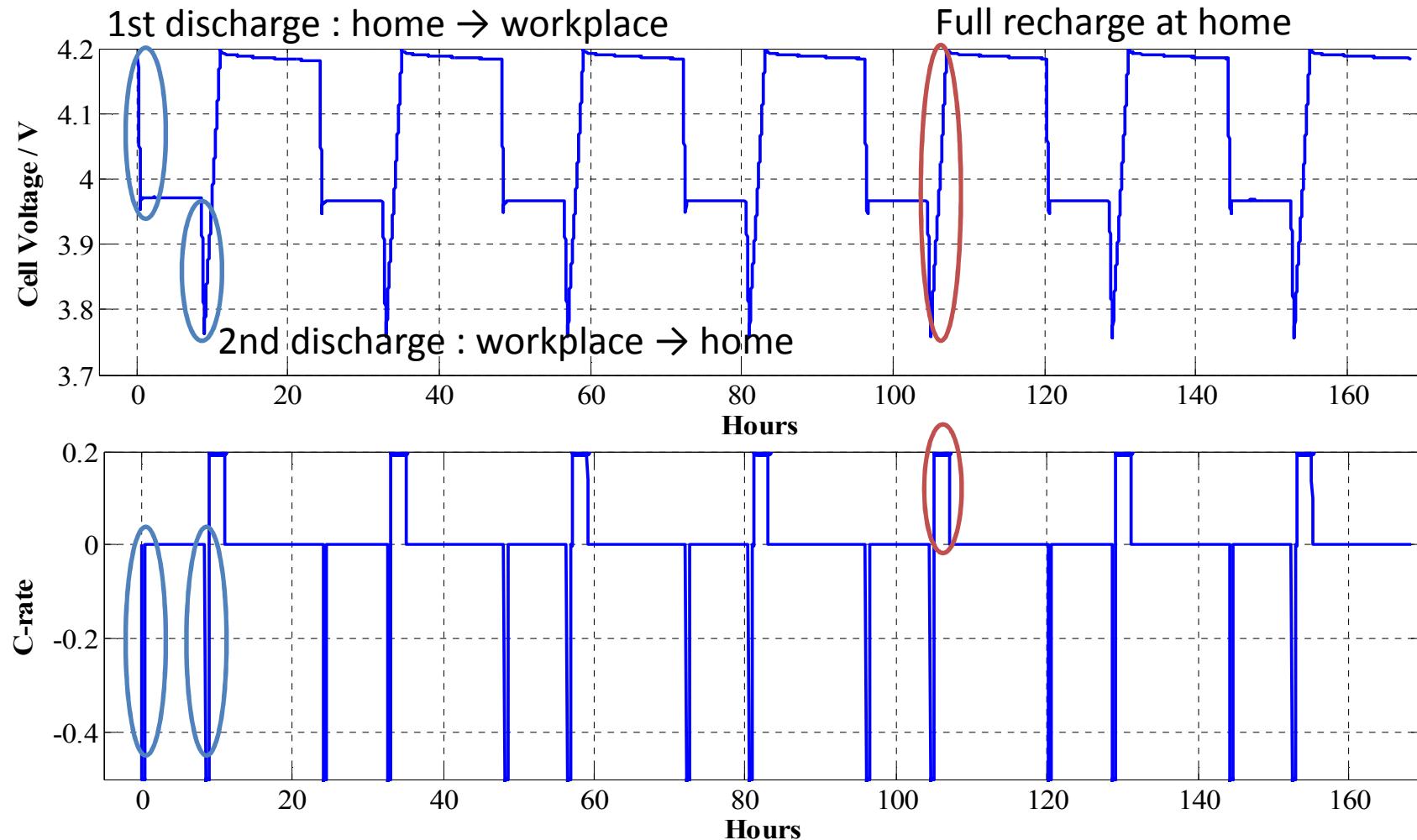
## 2. Cycling tests performed

- Mimic a cell usage of a pure EV (home  $\leftrightarrow$  workplace)
- Each trip = 20 % of DoD  $\rightarrow$  40% DoD /day



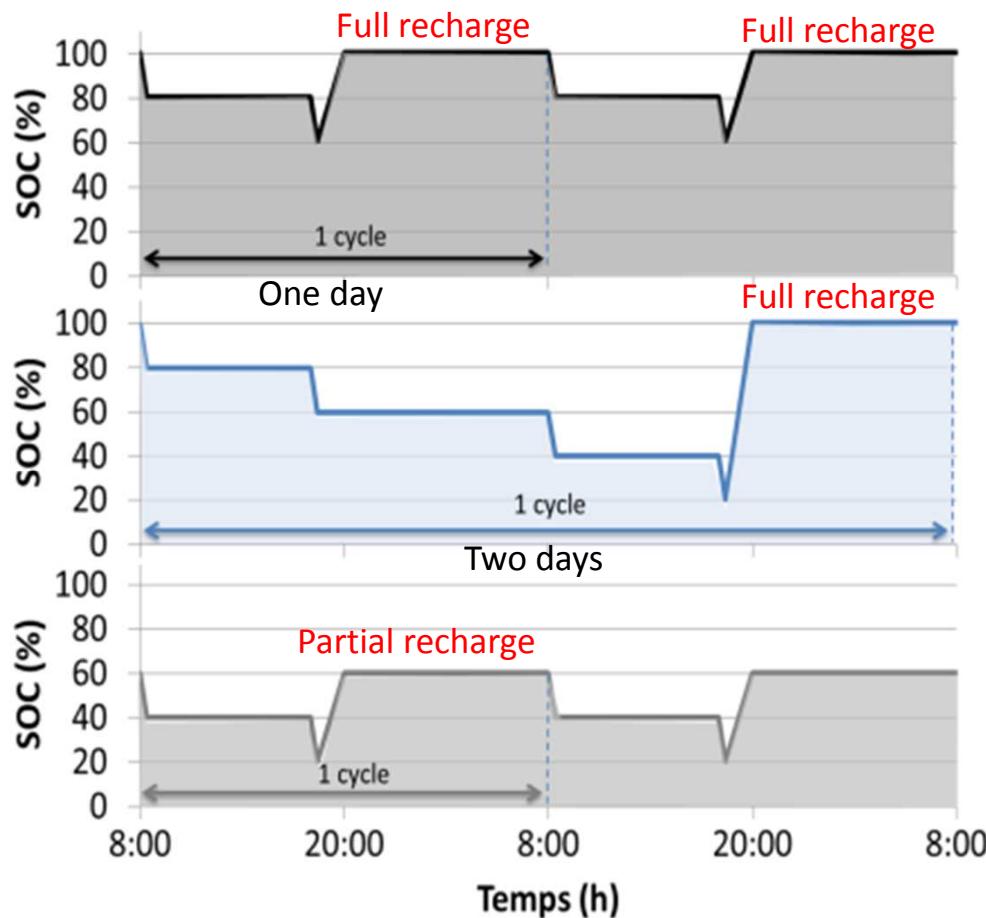
- Driving mode : C/2 constant current discharge
- Charging mode : C/5 constant current
- Profile repeated every day during 4 weeks

} Realistic current rates for EV

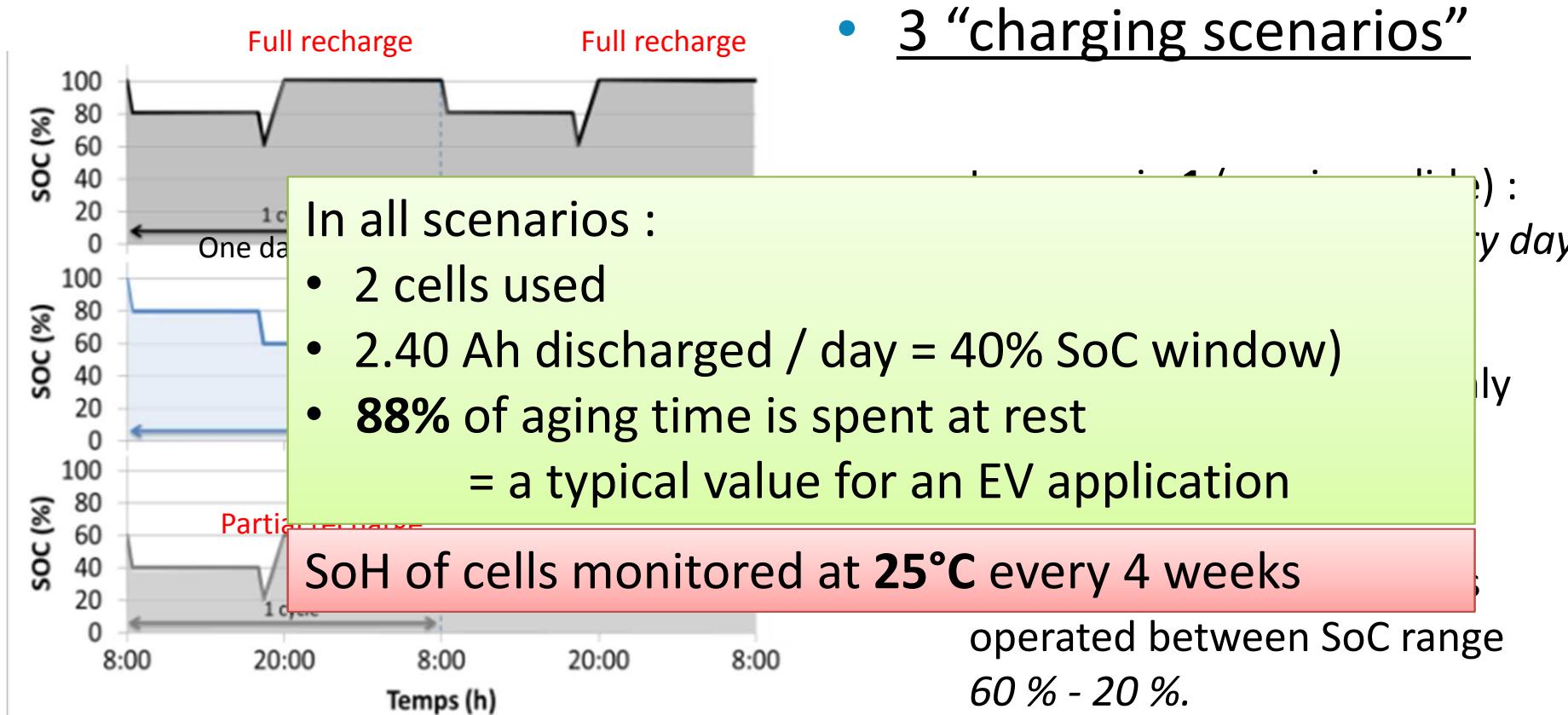


## Experiments : cycling aging

- 3 “charging scenarios”



- In scenario **1** (previous slide) : battery fully recharged *every day*, ,  $DoD=40\%$
- In scenario **2** : recharged only *when  $SoC=20\%$* ,  $DoD=80\%$
- In scenario **3** : the battery is operated between SoC range  $60\% - 20\%$ .



Test	Test description
Capacity test	<p>Fully discharge cell to 2.7V at <b>1C</b> rest 30 min</p> <p>Full charge to 4.2 V at <b>1C</b>, hold at 4.2V until the current drop below C/20 (CCCV)</p> <p>Discharge and charge three times</p> <p>Second and third capacity values are used to define the SoH</p>

- Periodicity:
  - Calendar aging experiments : every 8 weeks
  - Cycling aging experiments : every 4 weeks
- Other tests : Electrochemical Impedance Spectroscopy & relaxation test

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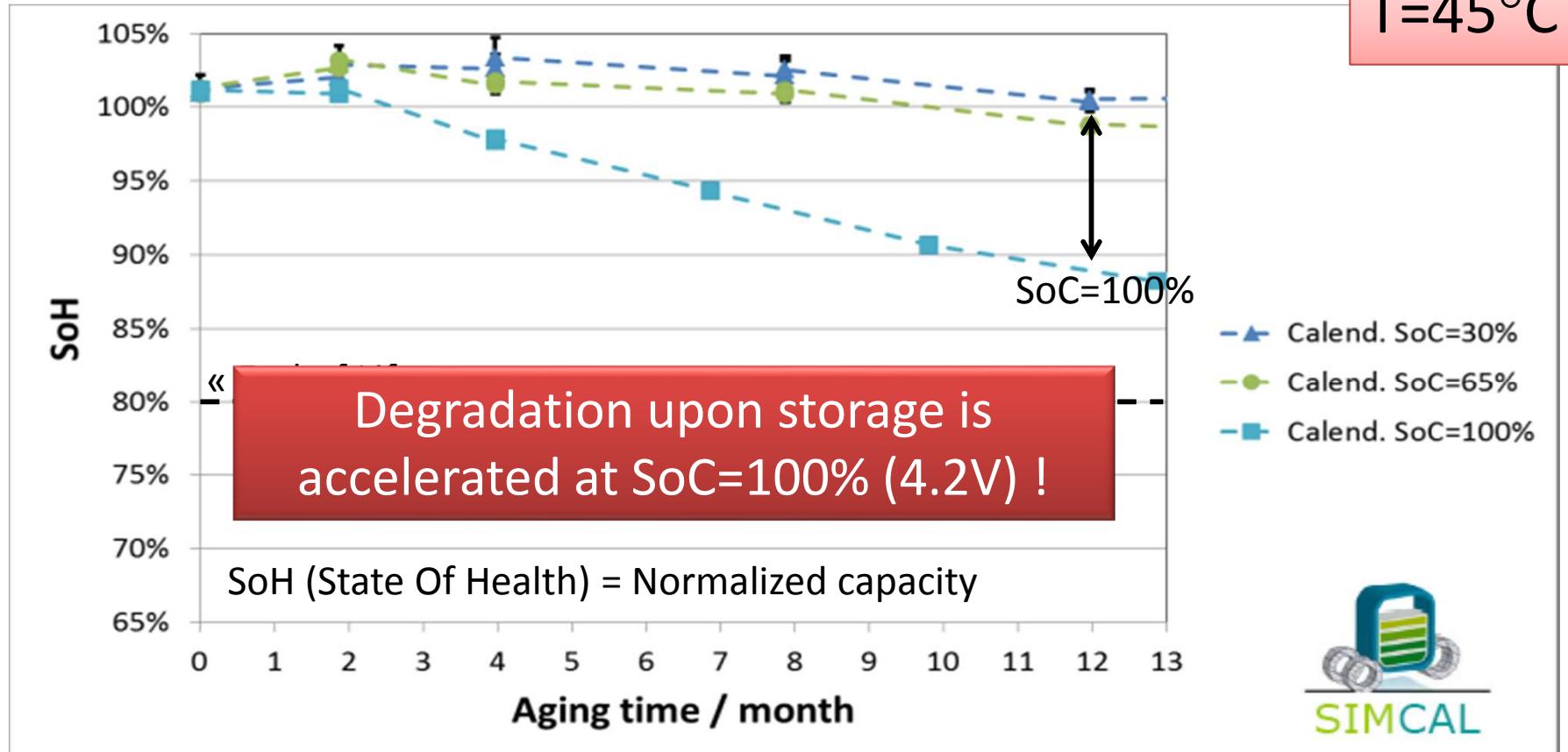


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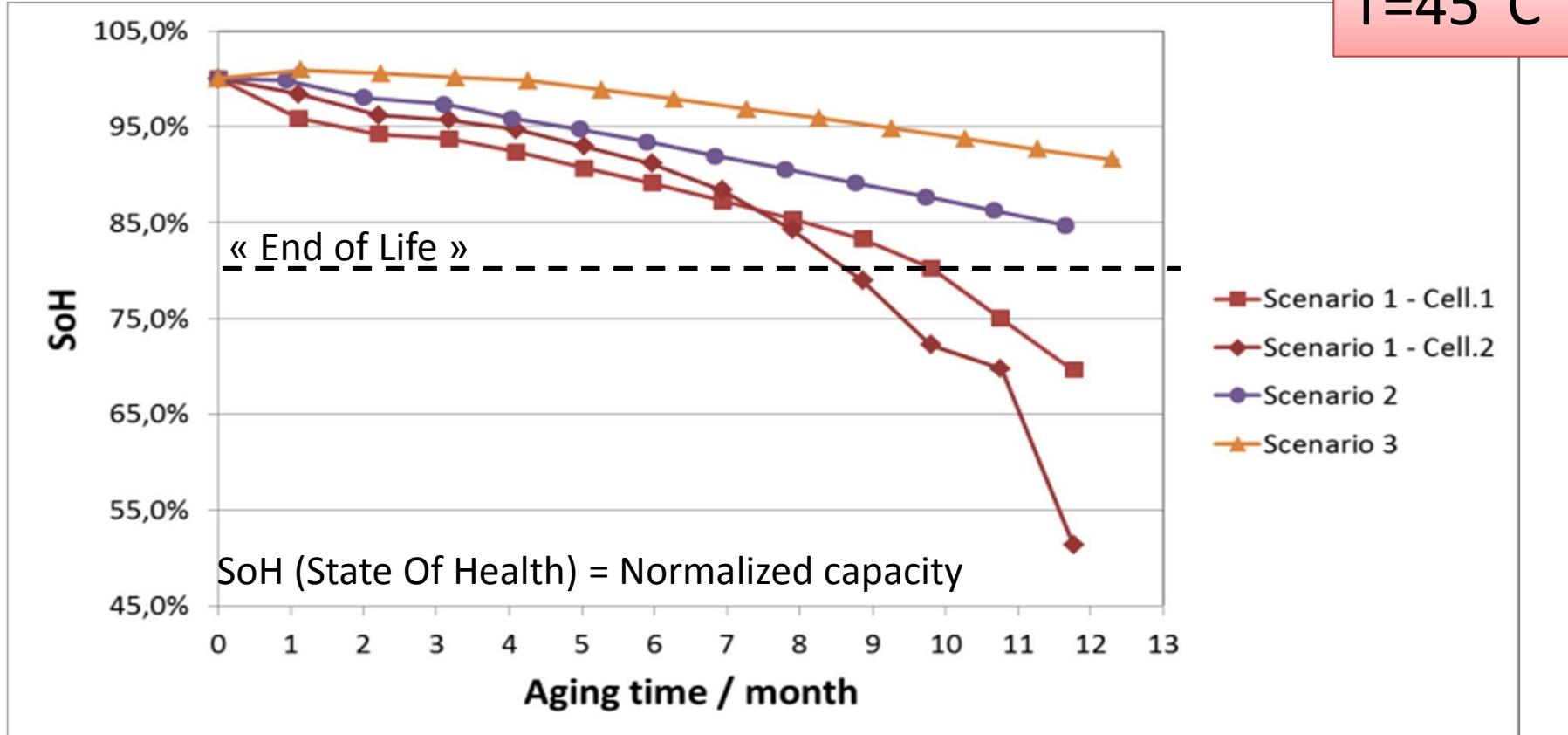
## 1. Calendar aging results

T=45°C



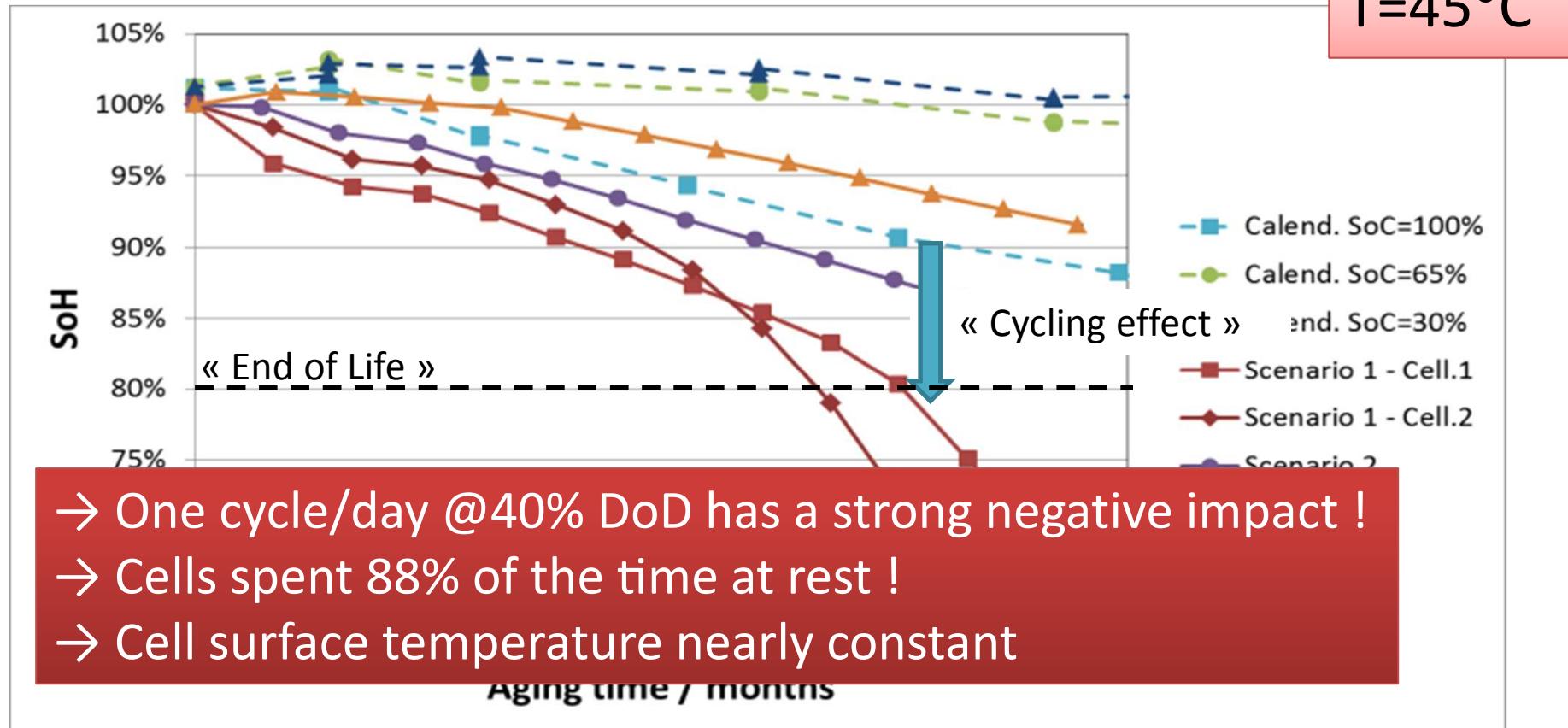
## 2. Cycling aging results

T=45°C



## 3. Cycling vs calendar aging

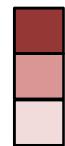
T=45°C



## Aging results : full cell capacity

Aging Test	End-Of-Life @ T=45°C (months)	Acceleration factor Cal. 45°C, SoC=100%
Calendar : 45°C, SoC=100%	20 <sup>m</sup>	1
<b>Scenario 1</b> <b>Charge every day to 100%</b>	<b>9<sup>m</sup></b>	<b>&gt; 2</b>
Scenario 2 Charge every <b>2 days</b> to 100%	15 <sup>e</sup>	1,3
Scenario 3 Charge every day to 60%	24 <sup>e</sup>	0,8

\* End-Of-Life : 20% of capacity loss



Capacity fade rate

m : measured

e : estimated by extrapolation of  
present test results

SoC management can be effective for preserving  
battery lifetime

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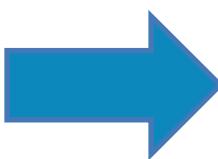
- Characterize degradation on electrode individually
- Disassembled in a glove box



Electrode sampling  
 $\varnothing = 14$  mm



Physical characterization (XRD<sup>1</sup>, SEM<sup>2</sup>, ICP-MS<sup>3</sup>)



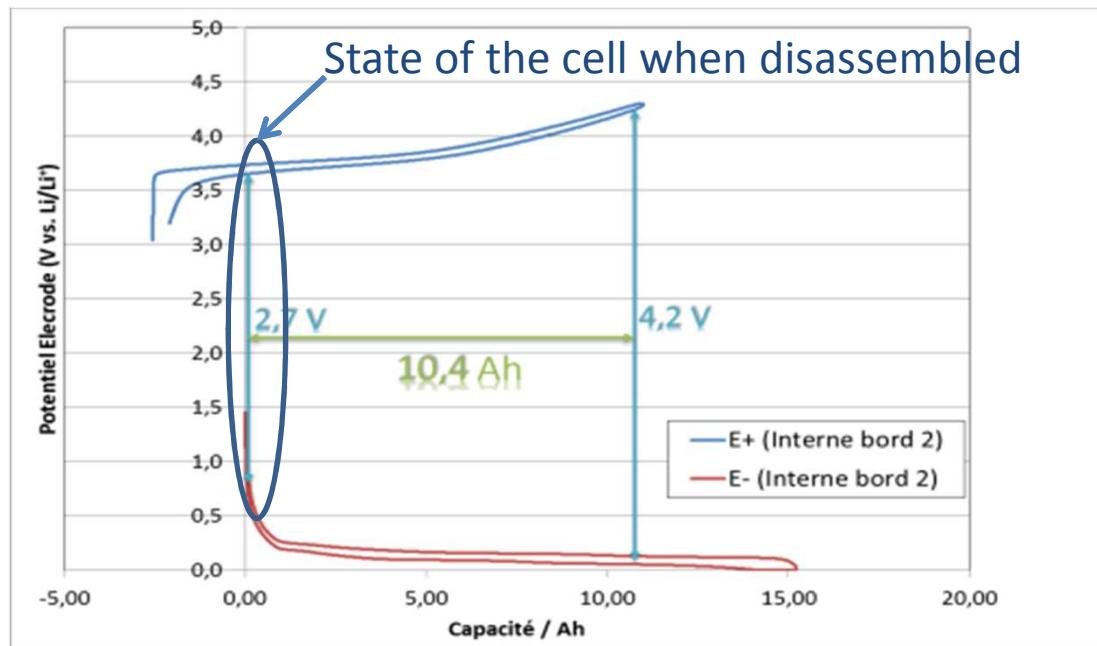
Electrochemical characterization

4 cells disassembled (1 ante-mortem + 3 from each cycling scenario)

- Cell #2 for scenario 1 (most aged cell)
- 9 XRD, 9 SEM, 8 ICP-MS analyses
- 50 assembled half-cells

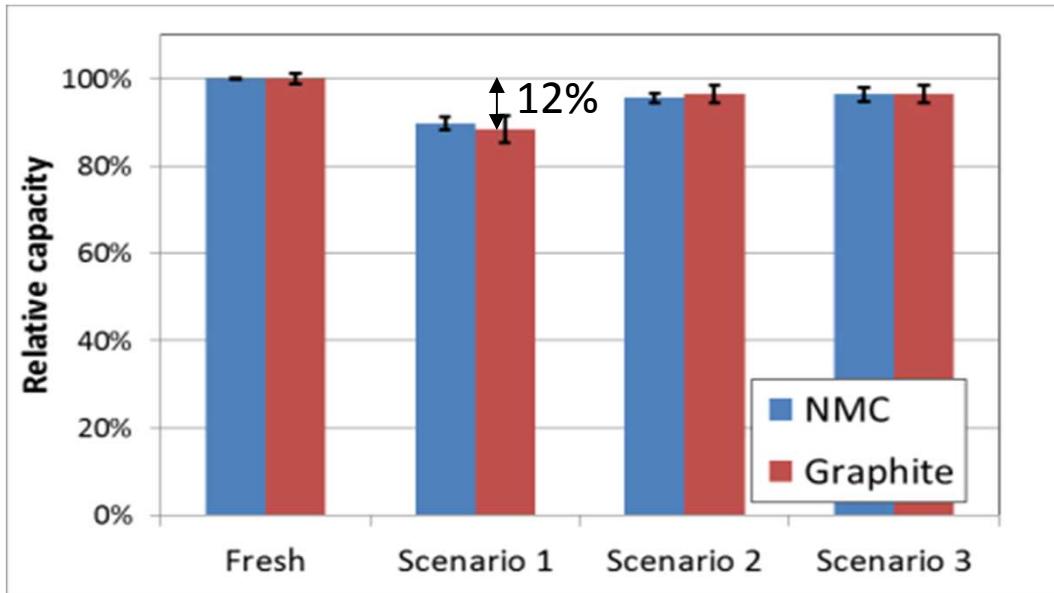
<sup>1</sup>X-Ray Diffraction, <sup>2</sup>Scanning Electron Microscopy, <sup>3</sup>Inductively Coupled Plasma Mass Spectrometry

- Electrochemical characterization of the electrodes
  1. Initial lithium content of the electrodes at discharge
  2. Capacities of the electrodes (4 cycles at C/10)

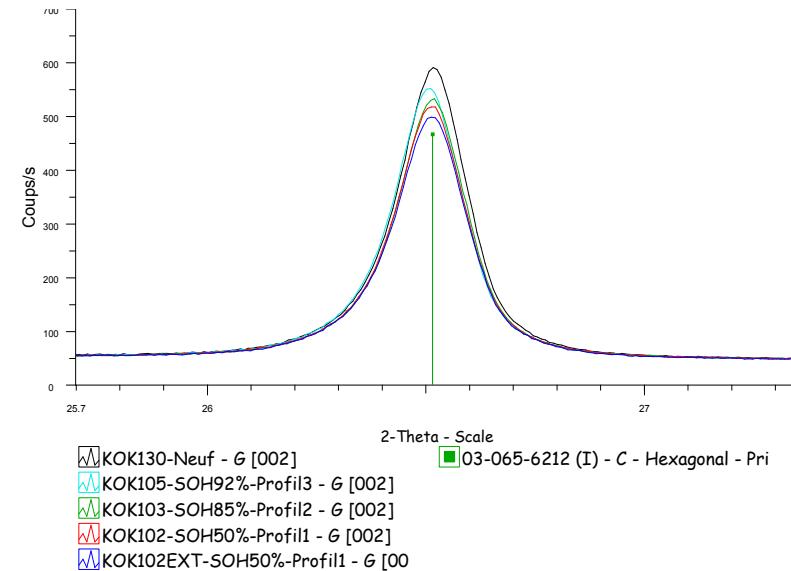


1. Negative electrode *is fully delithiated* when cell is disassembled
2. Initial lithiation of positive electrode provides cell internal imbalance

- Aging on electrode materials ?

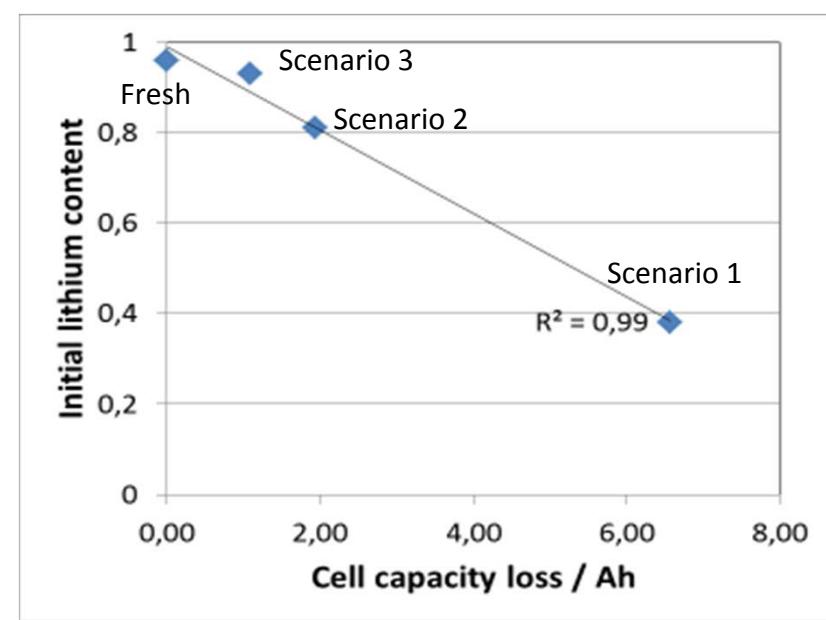
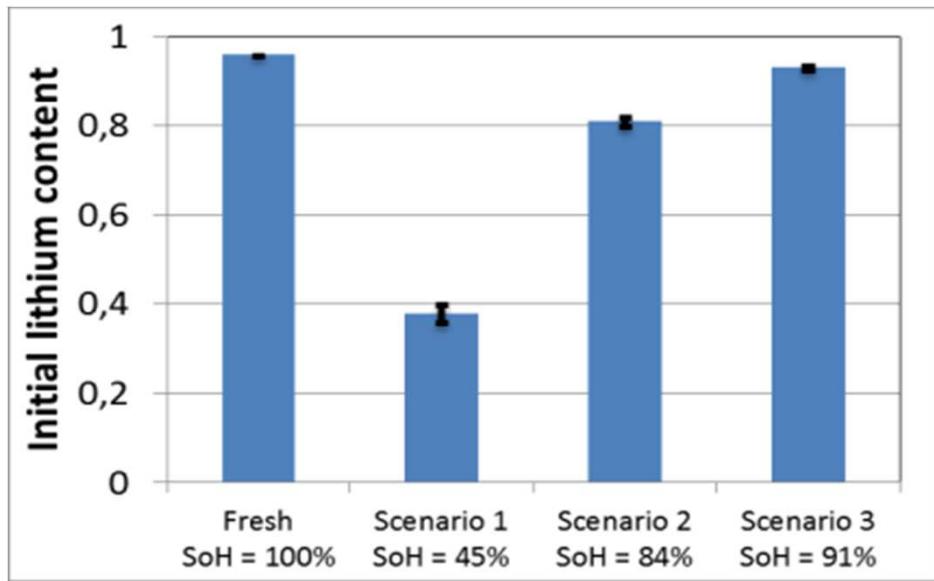


XRD on graphite electrode



- Loss of electrode capacities is rather limited: cannot explain cell capacity loss
- Confirmed by XRD analyses on negative electrode

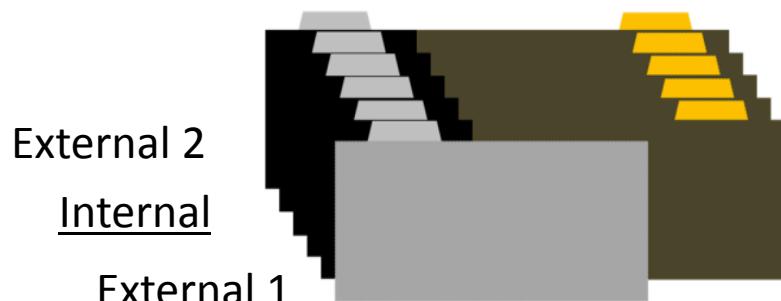
- Initial content of lithium in positive decreases with aging



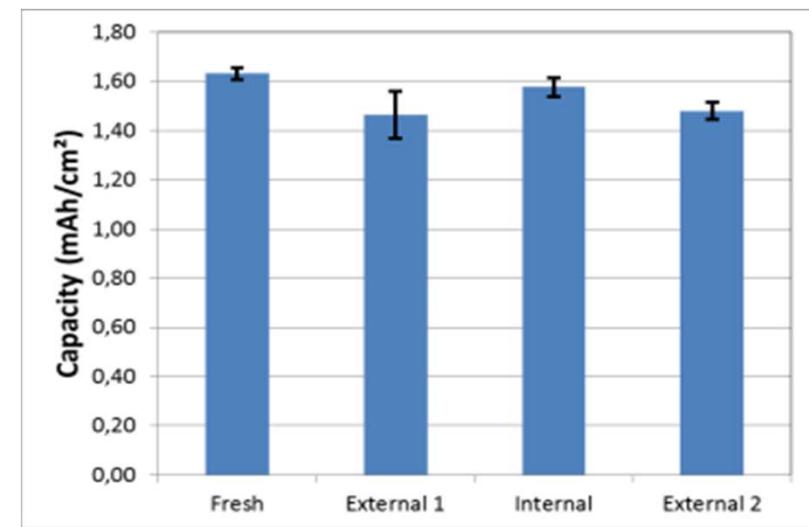
→ *increase of the cell imbalance...*

*whitch is well correlated with cell capacity*

- Inhomogeneous deterioration inside the cell was observed



Deposit observed on negative electrode plate located near the edge

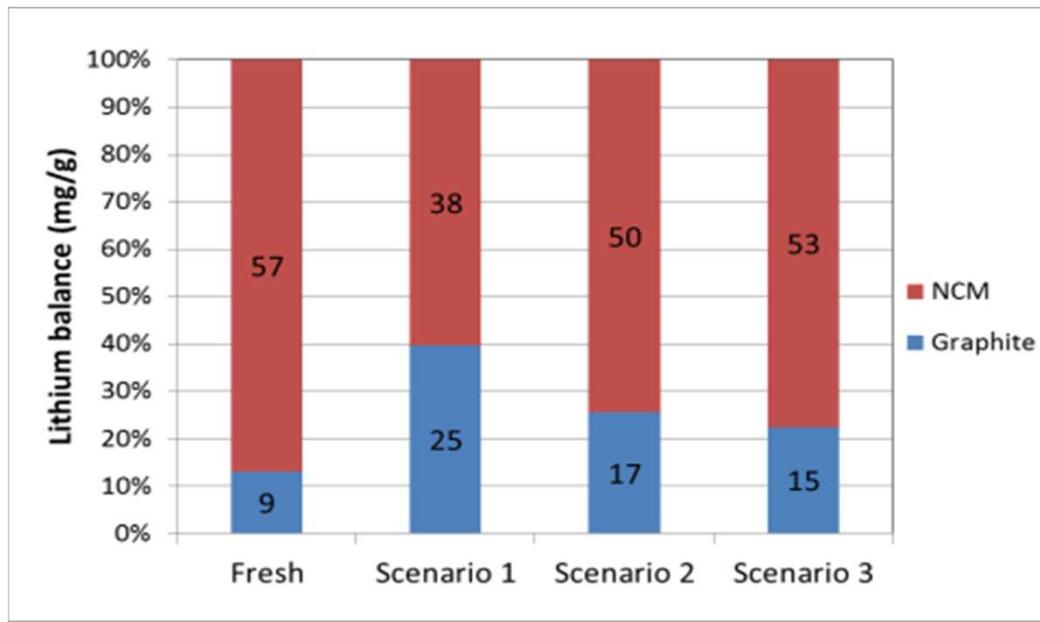


Cell from scenario 2

- Samples located on negative electrode plate near the edge exhibited **less capacity**

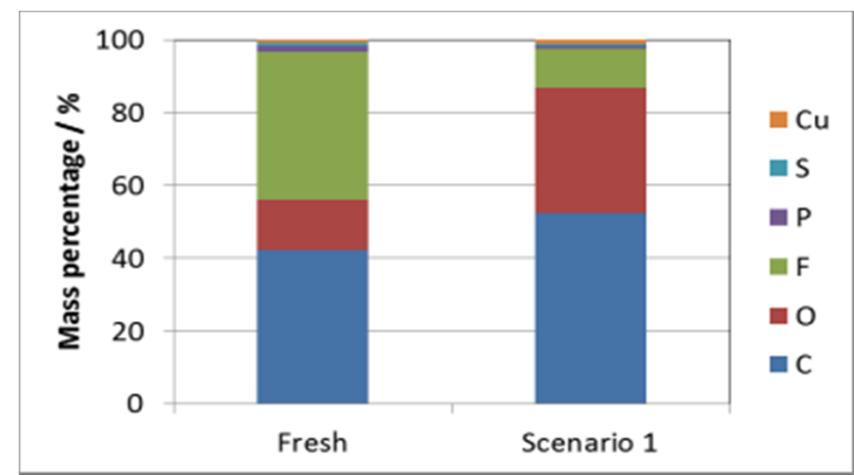
## Post-mortem characterization

- ICP-MS analysis



- ICP confirms loss of lithium in positive electrode
- Lithium content in negative electrode increases but is trapped into the SEI

- Energy dispersive analysis on negative electrode surface



- Increase of carbon and oxygen content

→ SEI growth on anode

- Compared to aging upon storage, a daily cycle representative of an EV usage increases substantially the aging rate
- **Charging only when necessary or/and by limiting charge voltage** can extend substantially cell lifetime
- For all tested aging conditions, analyses of individual electrodes conclude that loss of lithium cyclable is the main aging mechanism (**SEI continuous growth**)
- **Next step is to understand why SEI growth is accelerated by a daily use**

***Thank you for your attention!***

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